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Stereospecific ($2\pi+2\pi$) photocycloaddition of arylalkenes to pyrene via exciplex: formation of 1:1- and 2:1-cycloadducts

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Abstract

Irradiation of benzene solutions containing pyrene and electron-deficient arylalkenes such as E- and Zmethyl cinnamates afforded $(2\pi+2\pi)$ photocycloadducts including 1:2-cycloadduct in high yields in a stereospecific and *endo-selective manner*. Sandwich-type singlet exciplexes between pyrene and arylalkenes were proposed as reactive intermediates. \odot 2000 Elsevier Science Ltd. All rights reserved.

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Photocycloaddition of unsaturated compounds to aromatic rings has received considerable attention from synthetic and mechanistic viewpoints in the last three decades.1,2 Pyrene is a typical aromatic hydrocarbon which has wide-spread π -electrons and emits quite an intense blue monomer fluorescence and strong excimer fluorescence.³ Therefore, the photochemical and photophysical properties of pyrene monomer and/or excimer have been well investigated in the fields of the electron donor-acceptor interactions, the fluorescent probes, the chemosensors and so on.⁴ However, the photochemical reactivity of pyrene including the conversion of pyrene itself has been scarcely known.⁵ We now report the stereospecific and *endo-selective* ($2\pi+2\pi$) photocycloaddition of pyrene and pyrene derivatives with arylalkenes, giving 1:1- and 1:2-cycloadducts.

Irradiation of a benzene solution containing pyrene $(1, 0.02 \text{ mol/dm}^3)$ and an excess amount of E-methyl cinnamate $(t-2a, 0.1 \text{ mol/dm}^3)$ with a high-pressure Hg lamp through a Pyrex filter under an argon atmosphere for 1 h afforded two kinds of $(2\pi+2\pi)$ photocycloadducts *t*-3a and t -4a in a 25:1 ratio in high yields (80–90%) (Scheme 1). For prolonged irradiation, a 1:2-cycloadduct 5a was precipitated accompanying the formation of Z -methyl cinnamate $(c-2a)$ and trace

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amounts of other isomeric photocycloadducts c -3a and c -4a. Similar irradiation of 1 and c -2a stereospecifically afforded c -3a and c -4a in a 7:1 ratio in good yields.

These photocycloadducts were isolated by column chromatography on silica gel. Their structures were determined by their spectral properties. The ${}^{1}H$ NMR spectra of t -3a and c -3a showed the up-field shifted H_a protons at δ 6.38 and 6.65, which was due to an anisotropic effect of the phenyl ring. The methyl protons of t -4a and c -3a (δ 3.50 and 2.95) appeared at much higher fields than those of t -3a and c -4a (δ 3.65 and 3.64). Finally, the structure of t -3a was confirmed by X-ray crystallographic analysis.⁶ The structure of 5a, which was insoluble in chloroform and benzene, was also determined by the ¹H NMR spectrum in DMSO- d_6 at 130°C. The spectrum showed the symmetrical proton signals. The photoreaction of $t-3a$ with $t-2a$ afforded 5a exclusively. These results support the assigned structures for t -3a, t -4a, 5a, c -3a, and c -4a. It is noteworthy that the phenyl group lies at the *endo*-position in the major 1:1 adducts (*t*-3a and c-3a) and 1:2 adduct (5a) in spite of the predictable steric repulsion. Thus, it becomes clear that the photoreaction proceeds stereospecifically with stereoretention of the alkene used. Photochemical cycloreversion of t-3a hardly proceeded under the present reaction conditions, although it is well known that the cyclobutanes obtained by the photocycloaddition of alkenes to aromatic rings photocleaved to the starting substrates.^{1a,2a,7}

The photocycloaddition of 1 with *trans*- and *cis*-cinnamonitriles $(t-2b)$ and $c-2b$) similarly gave the 1:1 and 1:2 cycloadducts $(t-3b, t-4b, c-3b, c-4b, and 5b)$ in a stereospecific and *endo-selective* manner. The photoreaction of sterically hindered 2,7-di-tert-butylpyrene (6) with t -2a also gave the 1:1- and 1:2-cycloadducts $t-7$, $t-8$, and 9 in 61, 6, and 7% yields, respectively. Although

trans-stilbene (10a) and styrene did not react with 1 under the same reaction conditions, more electron-donating or electron-deficient stilbene derivatives such as $1,2$ -bis(4-methoxyphenyl)ethene $(10b)$ and 1,2-bis(4-cyanophenyl)ethene $(10c)$ stereospecifically afforded the corresponding $(2\pi+2\pi)$ photocycloadducts in 14 and 13% yields, respectively. Electron-deficient alkenes having no aryl substituent such as acrylonitrile and methyl acrylate, and electron-donating alkenes such as 2,3-dimethyl-2-butene and ethyl vinyl ether did not add to 1 under the same conditions.

From the mechanistic viewpoints, the triplet sensitized photoreaction and the fluorescence quenching experiments were carried out as follows: The photocycloaddition of t -2a or t -2b to 1 was not sensitized by triplet sensitizers such as benzophenone (69 kcal/mol) and Michler's ketone (61 kcal/mol). The monomer fluorescence of $1 (1 \times 10^{-5} \text{ mol/dm}^3)$ in benzene was efficiently quenched by t-2a,b, accompanying the appearance of a weak exciplex emission at longer wavelength (λ_{max}) ~450 nm) than the former emission of 1. The excimer fluorescence of 1 (1×10^{-3} mol/dm³, λ_{max}) \sim 500 nm) was also efficiently quenched by t-2a, accompanying the appearance of a weak exciplex emission at a shorter wavelength than the excimer emission. These results were reasonably elucidated by the singlet exciplex mechanism for the stereospecific and *endo-selective* photocycloaddition of arylalkenes to 1. Under the present reaction conditions ($\left[1\right] > 0.01$ mol/dm³), the primary process may be the formation of pyrene excimer 1_{2} ^{*} followed by the formation of exciplex 1_{1}^{1} . 2^{*} via exciplex (excimer) substitution.^{8,9} This exciplex produces the photocycloadduct efficiently. The *endo-selectivity* can be explained by the $\pi-\pi$ overlap interaction between 1 and the styryl chromophores of the arylalkenes via sandwich-type exciplexes as previously reported.1a,10 It is notable that the photoisomerization of arylalkenes was effectively suppressed by the presence of 1 under the reaction conditions, because the triplet energy of 1 is quite a bit lower than those of arylalkenes. Scope and detailed mechanism are now under investigation.

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- 6. Compound *t*-3a: mp 150°C; crystal data for *t*-3a: C₂₆H₂₀O₂, space group: $P2_1/a(\#14)$, $z=4$, $a=19.328(3)$, $b = 5.571(3), c = 19.527(4)$ \mathring{A} , $\beta = 114.83(1)$ °, $V = 1908(1)$ \mathring{A} ³, $R = 0.047$.
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- 9. The direct formation of the exciplex 1 [1...2]* from $11*$ and 2 is also a plausible pathway even in the high concentration of 1, because the formation of 1_{2} ^{*} from 11 ^{*} and 1 is reversible.
- 10. The excitation of bis(1-pyrenylmethyl) ether at 350 nm in benzene shows the excimer fluorescence at 507 nm almost exclusively even at low concentrations. This excimer fluorescence in benzene was efficiently quenched by t-2a accompanying the exciplex emission at shorter wavelength than the excimer emission. In addition, irradiation of bis(1-pyrenylmethyl) ether with $t-2a$ afforded a mixture of 1:1-cycloadducts although the structures were not decided. These results also support the exciplex mechanism, not the triplex mechanism.