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Stereospecific $(2\pi+2\pi)$ photocycloaddition of arylalkenes to pyrene via exciplex: formation of 1:1- and 2:1-cycloadducts

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Abstract

Irradiation of benzene solutions containing pyrene and electron-deficient arylalkenes such as *E*- and *Z*-methyl cinnamates afforded $(2\pi+2\pi)$ photocycloadducts including 1:2-cycloadduct in high yields in a stereospecific and *endo*-selective manner. Sandwich-type singlet exciplexes between pyrene and arylalkenes were proposed as reactive intermediates. © 2000 Elsevier Science Ltd. All rights reserved.

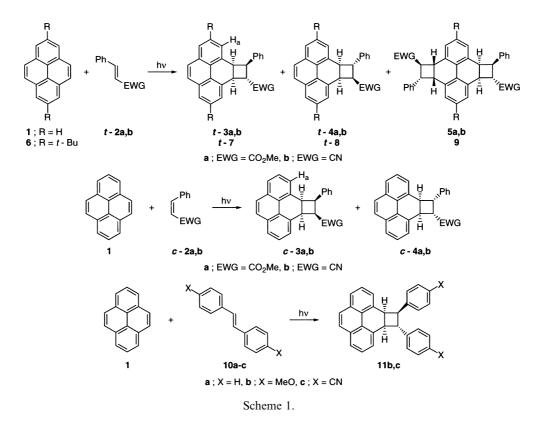
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Photocycloaddition of unsaturated compounds to aromatic rings has received considerable attention from synthetic and mechanistic viewpoints in the last three decades.^{1,2} Pyrene is a typical aromatic hydrocarbon which has wide-spread π -electrons and emits quite an intense blue monomer fluorescence and strong excimer fluorescence.³ Therefore, the photochemical and photophysical properties of pyrene monomer and/or excimer have been well investigated in the fields of the electron donor–acceptor interactions, the fluorescent probes, the chemosensors and so on.⁴ However, the photochemical reactivity of pyrene including the conversion of pyrene itself has been scarcely known.⁵ We now report the stereospecific and *endo*-selective ($2\pi+2\pi$) photocycloaddition of pyrene and pyrene derivatives with arylalkenes, giving 1:1- and 1:2-cycloadducts.

Irradiation of a benzene solution containing pyrene (1, 0.02 mol/dm³) and an excess amount of *E*-methyl cinnamate (*t*-2a, 0.1 mol/dm³) with a high-pressure Hg lamp through a Pyrex filter under an argon atmosphere for 1 h afforded two kinds of $(2\pi+2\pi)$ photocycloadducts *t*-3a and *t*-4a in a 25:1 ratio in high yields (80–90%) (Scheme 1). For prolonged irradiation, a 1:2-cyclo-adduct 5a was precipitated accompanying the formation of *Z*-methyl cinnamate (*c*-2a) and trace

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amounts of other isomeric photocycloadducts c-3a and c-4a. Similar irradiation of 1 and c-2a stereospecifically afforded c-3a and c-4a in a 7:1 ratio in good yields.



These photocycloadducts were isolated by column chromatography on silica gel. Their structures were determined by their spectral properties. The ¹H NMR spectra of *t*-3a and *c*-3a showed the up-field shifted H_a protons at δ 6.38 and 6.65, which was due to an anisotropic effect of the phenyl ring. The methyl protons of *t*-4a and *c*-3a (δ 3.50 and 2.95) appeared at much higher fields than those of *t*-3a and *c*-4a (δ 3.65 and 3.64). Finally, the structure of *t*-3a was confirmed by X-ray crystallographic analysis.⁶ The structure of 5a, which was insoluble in chloroform and benzene, was also determined by the ¹H NMR spectrum in DMSO-*d*₆ at 130°C. The spectrum showed the symmetrical proton signals. The photoreaction of *t*-3a, *t*-4a, 5a, *c*-3a, and *c*-4a. It is noteworthy that the phenyl group lies at the *endo*-position in the major 1:1 adducts (*t*-3a and *c*-3a) and 1:2 adduct (5a) in spite of the predictable steric repulsion. Thus, it becomes clear that the photoreaction proceeds stereospecifically with stereoretention of the alkene used. Photochemical cycloreversion of *t*-3a hardly proceeded under the present reaction conditions, although it is well known that the cyclobutanes obtained by the photocycloaddition of alkenes to aromatic rings photocleaved to the starting substrates.^{1a,2a,7}

The photocycloaddition of 1 with *trans*- and *cis*-cinnamonitriles (*t*-2**b** and *c*-2**b**) similarly gave the 1:1 and 1:2 cycloadducts (*t*-3**b**, *t*-4**b**, *c*-3**b**, *c*-4**b**, and 5**b**) in a stereospecific and *endo*-selective manner. The photoreaction of sterically hindered 2,7-di-*tert*-butylpyrene (6) with *t*-2**a** also gave the 1:1- and 1:2-cycloadducts *t*-7, *t*-8, and 9 in 61, 6, and 7% yields, respectively. Although

trans-stilbene (10a) and styrene did not react with 1 under the same reaction conditions, more electron-donating or electron-deficient stilbene derivatives such as 1,2-bis(4-methoxyphenyl)-ethene (10b) and 1,2-bis(4-cyanophenyl)ethene (10c) stereospecifically afforded the corresponding $(2\pi+2\pi)$ photocycloadducts in 14 and 13% yields, respectively. Electron-deficient alkenes having no aryl substituent such as acrylonitrile and methyl acrylate, and electron-donating alkenes such as 2,3-dimethyl-2-butene and ethyl vinyl ether did not add to 1 under the same conditions.

From the mechanistic viewpoints, the triplet sensitized photoreaction and the fluorescence quenching experiments were carried out as follows: The photocycloaddition of t-2a or t-2b to 1 was not sensitized by triplet sensitizers such as benzophenone (69 kcal/mol) and Michler's ketone (61 kcal/mol). The monomer fluorescence of 1 ($1 \times 10^{-5} \text{ mol/dm}^3$) in benzene was efficiently quenched by *t*-2a,b, accompanying the appearance of a weak exciplex emission at longer wavelength (λ_{max}) ~450 nm) than the former emission of 1. The excimer fluorescence of 1 (1×10^{-3} mol/dm³, λ_{max} \sim 500 nm) was also efficiently quenched by t-2a, accompanying the appearance of a weak exciplex emission at a shorter wavelength than the excimer emission. These results were reasonably elucidated by the singlet exciplex mechanism for the stereospecific and *endo*-selective photocycloaddition of arylalkenes to 1. Under the present reaction conditions ([1] $> 0.01 \text{ mol/dm}^3$), the primary process may be the formation of pyrene excimer ${}^{1}\mathbf{1}_{2}^{*}$ followed by the formation of exciplex ${}^{1}[\mathbf{1}\cdots\mathbf{2}]^{*}$ via exciplex (excimer) substitution.^{8,9} This exciplex produces the photocycloadduct efficiently. The *endo*-selectivity can be explained by the π - π overlap interaction between 1 and the styryl chromophores of the arylalkenes via sandwich-type exciplexes as previously reported.^{1a,10} It is notable that the photoisomerization of arylalkenes was effectively suppressed by the presence of 1 under the reaction conditions, because the triplet energy of 1 is quite a bit lower than those of arylalkenes. Scope and detailed mechanism are now under investigation.

Acknowledgements

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- 6. Compound *t*-3a: mp 150°C; crystal data for *t*-3a: $C_{26}H_{20}O_2$, space group: $P2_1/a(\#14)$, z=4, a=19.328(3), b=5.571(3), c=19.527(4) Å, $\beta=114.83(1)^\circ$, V=1908(1) Å³, R=0.047.
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- 9. The direct formation of the exciplex ${}^{1}[1\cdots 2]^{*}$ from ${}^{1}1^{*}$ and 2 is also a plausible pathway even in the high concentration of 1, because the formation of ${}^{1}1_{2}^{*}$ from ${}^{1}1^{*}$ and 1 is reversible.
- 10. The excitation of bis(1-pyrenylmethyl) ether at 350 nm in benzene shows the excimer fluorescence at 507 nm almost exclusively even at low concentrations. This excimer fluorescence in benzene was efficiently quenched by t-2a accompanying the exciplex emission at shorter wavelength than the excimer emission. In addition, irradiation of bis(1-pyrenylmethyl) ether with t-2a afforded a mixture of 1:1-cycloadducts although the structures were not decided. These results also support the exciplex mechanism, not the triplex mechanism.